

Chapter 7: Deformation & Strengthening Mechanisms

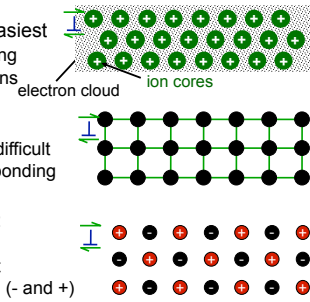
ISSUES TO ADDRESS...

- Why are the number of dislocations present greatest in metals?
- How are strength and dislocation motion related?
- Why does heating alter strength and other properties?

Chapter 7 - 1

Dislocations & Materials Classes

- Metals (Cu, Al):
Dislocation motion easiest
- non-directional bonding
- close-packed directions for slip
- Covalent Ceramics (Si, diamond): Motion difficult
- directional (angular) bonding
- Ionic Ceramics (NaCl):
Motion difficult
- need to avoid nearest neighbors of like sign (- and +)

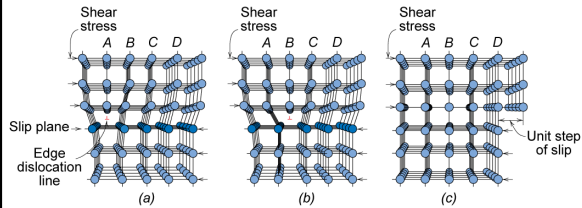


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Dislocation Motion

Dislocation motion & plastic deformation

- Metals - plastic deformation occurs by **slip** – an edge dislocation (extra half-plane of atoms) slides over adjacent plane half-planes of atoms.

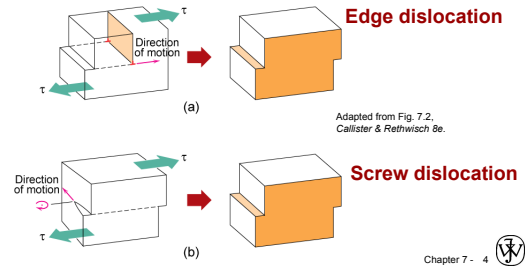


- If dislocations can't move, plastic deformation doesn't occur!

Adapted from Fig. 7.1, Callister & Rethwisch 8e.
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Dislocation Motion

- A dislocation moves along a **slip plane** in a **slip direction** perpendicular to the dislocation line
- The slip direction is the same as the **Burgers vector** direction



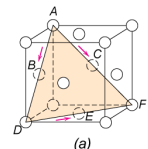
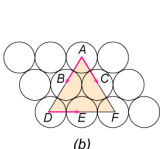
Adapted from Fig. 7.2, Callister & Rethwisch 8e.

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Deformation Mechanisms

Slip System

- Slip plane - plane on which easiest slippage occurs
 - Highest planar densities (and large interplanar spacings)
- Slip directions - directions of movement
 - Highest linear densities

Adapted from Fig. 7.6, Callister & Rethwisch 8e.

(a) (b)

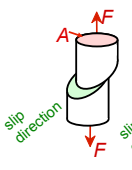
- FCC Slip occurs on {111} planes (close-packed) in <110> directions (close-packed)
=> total of 12 slip systems in FCC
- For BCC & HCP there are other slip systems.

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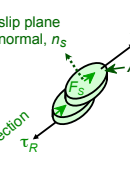
Stress and Dislocation Motion

- Resolved shear stress, τ_R
 - results from applied tensile stresses

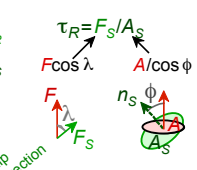
Applied tensile stress: $\sigma = F/A$



Resolved shear stress: $\tau_R = F_S/A_S$



Relation between σ and τ_R



$\tau_R = \sigma \cos \lambda \cos \phi$
 $\lambda = \text{angle_between_slip \& stress}$
 $\phi = \text{angle_normal_to_slip \& stress_direction}$

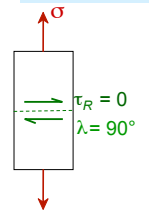
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Critical Resolved Shear Stress

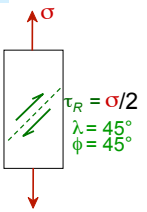
- Condition for dislocation motion: $\tau_R > \tau_{CRSS}$
- Ease of dislocation motion depends on crystallographic orientation

typically 10^{-4} GPa to 10^{-2} GPa

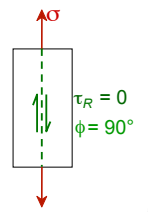
$\tau_R = \sigma \cos \lambda \cos \phi$



$\tau_R = 0$
 $\lambda = 90^\circ$



$\tau_R = \sigma/2$
 $\lambda = 45^\circ$
 $\phi = 45^\circ$

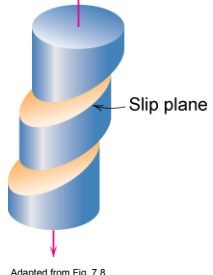


$\tau_R = 0$
 $\phi = 90^\circ$

τ maximum at $\lambda = \phi = 45^\circ$


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Single Crystal Slip



Direction of force

Slip plane



Adapted from Fig. 7.9, Callister & Rethwisch 8e.

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Slip Motion in Polycrystals

- Polycrystals stronger than single crystals – grain boundaries are barriers to dislocation motion.
- Slip planes & directions (λ , ϕ) change from one grain to another.
- τ_R will vary from one grain to another.
- The grain with the largest τ_R yields first.
- Other (less favorably oriented) grains yield later.

Adapted from Fig. 7.10, Callister & Rethwisch 8e. (Fig. 7.10 is courtesy of C. Brady, National Bureau of Standards [now the National Institute of Standards and Technology, Gaithersburg, MD].)

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Anisotropy in σ_y

- Can be induced by rolling a polycrystalline metal
 - before rolling
 - after rolling

Adapted from Fig. 7.11, Callister & Rethwisch 8e. (Fig. 7.11 is from W.G. Moffatt, G.W. Pearsall, and J. Wulff, *The Structure and Properties of Materials*, Vol. I, Structure, p. 140, John Wiley and Sons, New York, 1964.)

- isotropic since grains are equiaxed & randomly oriented.
- anisotropic since rolling affects grain orientation and shape.

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Anisotropy in Deformation

- Cylinder of tantalum machined from a rolled plate:
- Fire cylinder at a target.
- Deformed cylinder

Photos courtesy of G.T. Gray III, Los Alamos National Labs. Used with permission.

- The noncircular end view shows anisotropic deformation of rolled material.

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Four Strategies for Strengthening: 1: Reduce Grain Size

- Grain boundaries are barriers to slip.
- Barrier "strength" increases with increasing angle of misorientation.
- Smaller grain size: more barriers to slip.

Adapted from Fig. 7.14, Callister & Rethwisch 8e. (Fig. 7.14 is from *A Textbook of Materials Technology*, by Van Vlack, Pearson Education, Inc., Upper Saddle River, NJ.)

- Hall-Petch Equation: $\sigma_{yield} = \sigma_o + k_y d^{-1/2}$

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Four Strategies for Strengthening: 2: Form Solid Solutions

- Impurity atoms distort the lattice & generate lattice strains.
- These strains can act as barriers to dislocation motion.
- Smaller substitutional impurity
- Larger substitutional impurity

Impurity generates local stress at **A** and **B** that opposes dislocation motion to the right.

Impurity generates local stress at **C** and **D** that opposes dislocation motion to the right.

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Lattice Strains Around Dislocations

Adapted from Fig. 7.4, Callister & Rethwisch 8e.

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Strengthening by Solid Solution Alloying

- Small impurities tend to concentrate at dislocations (regions of compressive strains) - partial cancellation of dislocation compressive strains and impurity atom tensile strains
- Reduce mobility of dislocations and increase strength

(a)

(b)

Adapted from Fig. 7.17, Callister & Rethwisch 8e.

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Strengthening by Solid Solution Alloying

- Large impurities tend to concentrate at dislocations (regions of tensile strains)

(a)

(b)

Adapted from Fig. 7.18, Callister & Rethwisch 8e.

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VMSE Solid-Solution Strengthening Tutorial

Solid-Solution Strengthening

This module allows you to observe the mechanism of solid solution strengthening by viewing animations, which are accompanied with voice-overs.

Pure

Larger

Smaller

Interstitial

Please be patient. This may take a few minutes to load depending on your connection speed.

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Ex: Solid Solution Strengthening in Copper

- Tensile strength & yield strength increase with wt% Ni.

Tensile strength (MPa)

wt.% Ni, (Concentration C)

Yield strength (MPa)

wt.% Ni, (Concentration C)

Adapted from Fig. 7.16(a) and (b), Callister & Rethwisch 8e.

- Empirical relation: $\sigma_y \sim C^{1/2}$
- Alloying increases σ_y and *TS*.

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Four Strategies for Strengthening: 3: Precipitation Strengthening

- Hard precipitates are difficult to shear.
Ex: Ceramics in metals (SiC in Iron or Aluminum).

Side View

Top View

Large shear stress needed to move dislocation toward precipitate and shear it.

Dislocation "advances" but precipitates act as "pinning" sites with spacing *S*.

- Result: $\sigma_y \sim \frac{1}{S}$

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Application: Precipitation Strengthening

- Internal wing structure on Boeing 767

Adapted from chapter-opening photograph, Chapter 11, Callister & Rethwisch 3e, (courtesy of G.H. Narayanan and A.G. Miller, Boeing Commercial Airplane Company.)

- Aluminum is strengthened with precipitates formed by alloying.

Adapted from Fig. 11.26, Callister & Rethwisch 8e. (Fig. 11.26 is courtesy of G.H. Narayanan and A.G. Miller, Boeing Commercial Airplane Company.)

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Four Strategies for Strengthening: 4: Cold Work (Strain Hardening)

- Deformation at room temperature (for most metals).
- Common forming operations reduce the cross-sectional area:

-Forging

-Rolling

-Drawing

-Extrusion

Adapted from Fig. 11.8, Callister & Rethwisch 8e.

$$\%CW = \frac{A_0 - A_d}{A_0} \times 100$$

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Dislocation Structures Change During Cold Working

- Dislocation structure in Ti after cold working.

- Dislocations entangle with one another during cold work.
- Dislocation motion becomes more difficult.

Fig. 4.6, Callister & Rethwisch 8e. (Fig. 4.6 is courtesy of M.R. Pichia, Michigan Technological University.)

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Dislocation Density Increases During Cold Working

Dislocation density = $\frac{\text{total dislocation length}}{\text{unit volume}}$

- Carefully grown single crystals
→ ca. 10^3 mm^{-2}
- Deforming sample increases density
→ $10^9\text{-}10^{10} \text{ mm}^{-2}$
- Heat treatment reduces density
→ $10^5\text{-}10^6 \text{ mm}^{-2}$

- Yield stress increases as ρ_d increases:

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Impact of Cold Work

As cold work is increased

- Yield strength (σ_y) increases.
- Tensile strength (TS) increases.
- Ductility ($\%EL$ or $\%AR$) decreases.

Adapted from Fig. 7.20, Callister & Rethwisch 8e.

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Mechanical Property Alterations Due to Cold Working

- What are the values of yield strength, tensile strength & ductility after cold working Cu?

$$\%CW = \frac{\pi D_0^2 - \pi D_d^2}{\pi D_0^2} \times 100 = \frac{D_0^2 - D_d^2}{D_0^2} \times 100$$

$$\%CW = \frac{(15.2 \text{ mm})^2 - (12.2 \text{ mm})^2}{(15.2 \text{ mm})^2} \times 100 = 35.6\%$$

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Mechanical Property Alterations Due to Cold Working

- What are the values of yield strength, tensile strength & ductility for Cu for %CW = 35.6%?

$\sigma_y = 300 \text{ MPa}$ $TS = 340 \text{ MPa}$ $\%EL = 7\%$

Adapted from Fig. 7.19, Callister & Rethwisch 8e. (Fig. 7.19 is adapted from Metals Handbook: Properties and Selection: Iron and Steels, Vol. 1, 9th ed., B. Bardes (Ed.), American Society for Metals, 1978, p. 226; and Metals Handbook: Properties and Selection: Nonferrous Alloys and Pure Metals, Vol. 2, 9th ed., H. Baker (Managing Ed.), American Society for Metals, 1979, p. 276 and 327.)

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Effect of Heat Treating After Cold Working

- 1 hour treatment at T_{anneal} ... decreases TS and increases %EL.
- Effects of cold work are nullified!

- Three Annealing stages:
 1. Recovery
 2. Recrystallization
 3. Grain Growth

Adapted from Fig. 7.22, Callister & Rethwisch 8e. (Fig. 7.22 is adapted from G. Sachs and K.R. van Horn, Practical Metallurgy, Applied Metallurgy, and the Industrial Processing of Ferrous and Nonferrous Metals and Alloys, American Society for Metals, 1940, p. 139.)

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Three Stages During Heat Treatment: 1. Recovery

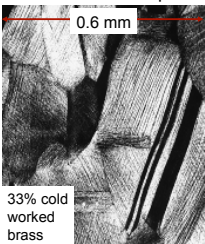
Reduction of dislocation density by annihilation.

- Scenario 1: Results from diffusion. Diagram shows dislocations annihilating to form a perfect atomic plane.
- Scenario 2:
 1. dislocation blocked; can't move to the right
 2. grey atoms leave by vacancy diffusion allowing disl. to "climb"
 3. "Climbed" disl. can now move on new slip plane
 4. opposite dislocations meet and annihilate

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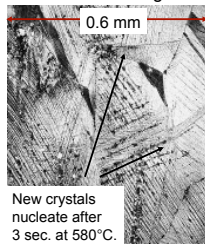
Three Stages During Heat Treatment: 2. Recrystallization

- New grains are formed that:
 - have low dislocation densities
 - are small in size
 - consume and replace parent cold-worked grains.



0.6 mm

33% cold worked brass



0.6 mm

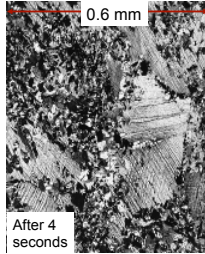
New crystals nucleate after 3 sec. at 580°C.

Adapted from Fig. 7.21(a),(b), Callister & Rethwisch 8e. (Fig. 7.21(a),(b) are courtesy of J.E. Burke, General Electric Company.)

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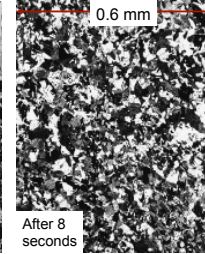
As Recrystallization Continues...

- All cold-worked grains are eventually consumed/replaced.



0.6 mm

After 4 seconds



0.6 mm

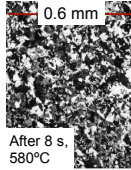
After 8 seconds

Adapted from Fig. 7.21(c),(d), Callister & Rethwisch 8e. (Fig. 7.21(c),(d) are courtesy of J.E. Burke, General Electric Company.)

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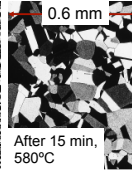
Three Stages During Heat Treatment: 3. Grain Growth

- At longer times, average grain size increases.
 - Small grains shrink (and ultimately disappear)
 - Large grains continue to grow



0.6 mm

After 8 s, 580°C



0.6 mm

After 15 min, 580°C

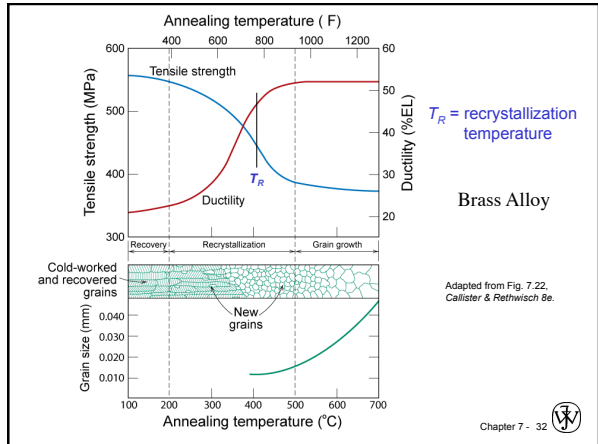
Adapted from Fig. 7.21(d),(e), Callister & Rethwisch 8e. (Fig. 7.21(d),(e) are courtesy of J.E. Burke, General Electric Company.)

Empirical Relation:

$$d^n - d_o^n = Kt$$

exponent typ. ~ 2
 grain diam. at time t. d^n
 coefficient dependent on material and T.
 elapsed time t

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Recrystallization Temperature

T_R = recrystallization temperature = temperature at which recrystallization just reaches completion in 1 h.

$$0.3T_m < T_R < 0.6T_m$$

For a specific metal/alloy, T_R depends on:

- %CW -- T_R decreases with increasing %CW
- Purity of metal -- T_R decreases with increasing purity

Recrystallization & Melting

Table 7.2 Recrystallization and Melting Temperatures for Various Metals and Alloys

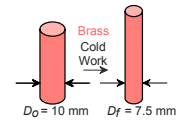
Metal	Recrystallization Temperature		Melting Temperature	
	°C	°F	°C	°F
Lead	-4	25	327	620
Tin	-4	25	232	450
Zinc	10	50	420	788
Aluminum (99.999 wt%)	80	176	660	1220
Copper (99.999 wt%)	120	250	1085	1985
Brass (60 Cu-40 Zn)	475	887	900	1652
Nickel (99.99 wt%)	370	700	1455	2651
Iron	450	840	1538	2800
Tungsten	1200	2200	3410	6170

Diameter Reduction Procedure - Problem

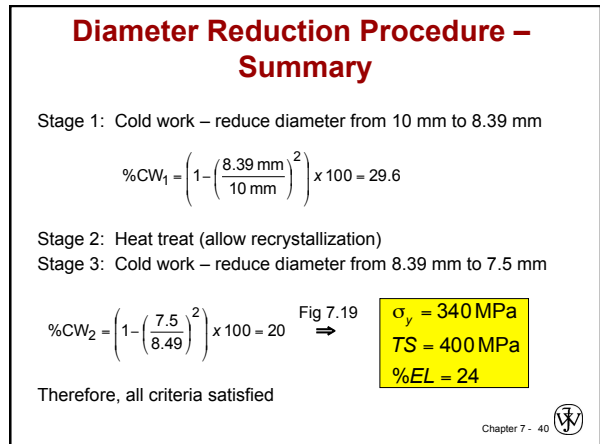
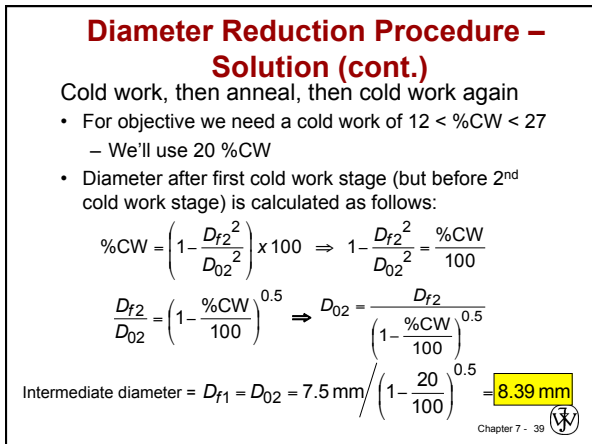
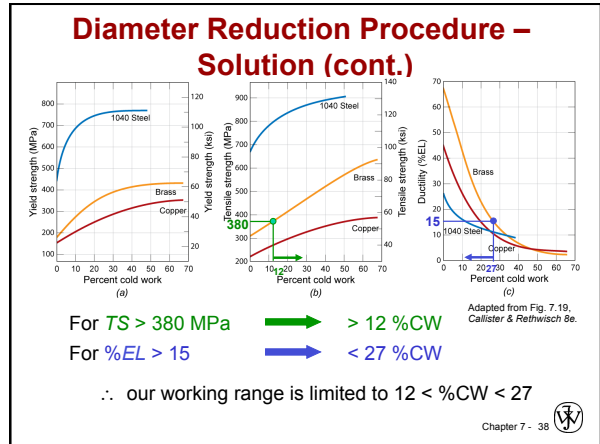
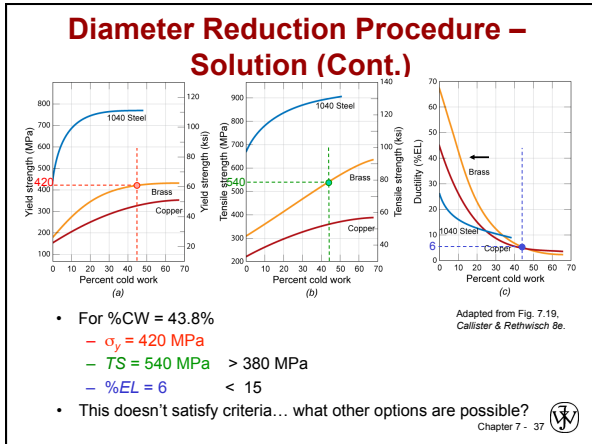
A cylindrical rod of brass originally 10 mm (0.39 in) in diameter is to be cold worked by drawing. The circular cross section will be maintained during deformation. A cold-worked tensile strength in excess of 380 MPa (55,000 psi) and a ductility of at least 15 %EL are desired. Furthermore, the final diameter must be 7.5 mm (0.30 in). Explain how this may be accomplished.

Diameter Reduction Procedure - Solution

What are the consequences of directly drawing to the final diameter?



$$\begin{aligned} \%CW &= \left(\frac{A_o - A_f}{A_o} \right) \times 100 = \left(1 - \frac{A_f}{A_o} \right) \times 100 \\ &= \left(1 - \frac{\pi D_f^2 / 4}{\pi D_o^2 / 4} \right) \times 100 = \left(1 - \left(\frac{7.5}{10} \right)^2 \right) \times 100 = 43.8\% \end{aligned}$$



Cold Working vs. Hot Working

- **Hot working** → deformation above T_R
- **Cold working** → deformation below T_R

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Grain Size Influences Properties

- **Metals having small grains** – relatively strong and tough at low temperatures
- **Metals having large grains** – good creep resistance at relatively high temperatures

Chapter 7 -



Summary

- Dislocations are observed primarily in metals and alloys.
- Strength is increased by making dislocation motion difficult.
- Strength of metals may be increased by:
 - decreasing grain size
 - **solid solution strengthening**
 - **precipitate hardening**
 - **cold working**
- A cold-worked metal that is heat treated may experience recovery, recrystallization, and grain growth – its properties will be altered.

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ASSIGNMENT/DUE DATE

Reading: Chapters 7 & 8/2-28-12

Problems: 7.5, 7.22, 7.27, 7.D7/3-1-12

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