

Physics 112

Introduction and Chapter 15

Formalities

- * My name is Dr. Jared Workman
- * Before you leave try and make a friend and get contact info
- * What's this course about? Let's go to the webpage

Review

- * What I expect you to know
 - * Vectors
 - * Newton's Laws
 - * Algebra
- * I will not be covering every step of every problem

Assistance

- * Me
- * Friends
 - * I encourage you to work together
- * Tutoring Center
- * Links on my site

One important Caveat

- * If you hand in a homework problem with no work done it will be graded a zero**

Why do you care?

- * This stuff forms the basis for
 - * Power
 - * Medicine
 - * Optics
 - * All modern electronics

Let's start with Chapter 15

- * We will not cover Gauss's Law or the Van Der Graaff Generator. In general your study guide is essentially what I tell you not to cover

Forces

- * You've seen many, normal, gravitational, frictional, tension, bouyant, etc
- * There are really only 4
 - * Strong
 - * Weak
 - * Electromagnetic
 - * Gravitational

- * All phenomenon are manifestations of these four fundamental forces
- * Let's start with the electromagnetic force
- * The electromagnetic force is 10^{36} times more powerful between two particles than the gravitational force
- * Why don't we see it in everyday life?
- * Why does so much energy go into generating electricity?

EM force

- * There are two types of charges, positive and negative
- * Like charges repel and unlike charges repel
 - * protons, electron, neutrons
- * The SI unit of force is the coulomb and is given by $e = 1.6 \times 10^{-19} \text{ C}$
- * Protons = $+e$, electrons = $-e$, neutrons = $0e$
- * Charge is a conserved quantity
- * Charge is a quantized quantity, it always comes in units of $\pm ne$, where n is an integer

Let's stop for a moment

- * How do nuclei stay together?

Demo

- * Rub a glass rod with silk
- * Electrons are transferred to the silk, leaves an equivalent number of protons on the rod
- * There is a force clearly operating here

2 basic types of materials

Conductors

- Conductors are materials in which the electric charges move freely in response to an electric force.
 - Copper, aluminum and silver are good conductors.
 - When a conductor is charged in a small region, the charge readily distributes itself over the entire surface of the material.

Conductors in Electrostatic Equilibrium

- When no net motion of charge occurs within a conductor, the conductor is said to be in **electrostatic equilibrium**.
- An isolated conductor has the following properties:
 - The electric field is zero everywhere inside the conducting material.
 - Any excess charge on an isolated conductor resides entirely on its surface.

Section 15.6

Properties, Cont.

- The electric field just outside a charged conductor is perpendicular to the conductor's surface.
- On an irregularly shaped conductor, the charge accumulates at locations where the radius of curvature of the surface is smallest (that is, at sharp points).

Section 15.6

Insulators

- Insulators are materials in which electric charges do not move freely.
 - Glass and rubber are examples of insulators.
 - When insulators are charged by rubbing, only the rubbed area becomes charged.
 - There is no tendency for the charge to move into other regions of the material.

Also

Semiconductors

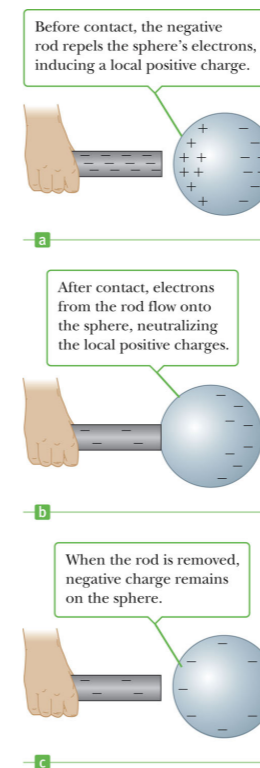
- The characteristics of semiconductors are between those of insulators and conductors.
- Silicon and germanium are examples of semiconductors.

How to impart a net charge to an object

- * Conduction - touch
- * Induction - no touch

Charging by Conduction

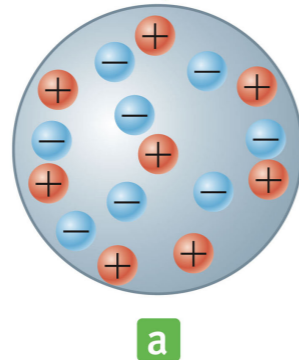
- A charged object (the rod) is placed in contact with another object (the sphere).
- Some electrons on the rod can move to the sphere.
- When the rod is removed, the sphere is left with a net negative charge.
- The object being charged is always left with a charge having the same sign as the object doing the charging.



Charging by Induction

- When an object is connected to a conducting wire or pipe buried in the earth, it is said to be grounded.
- A neutral sphere has equal number of electrons and protons.

The neutral sphere has equal numbers of positive and negative charges.

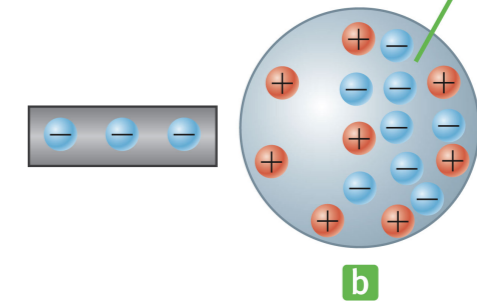


Section 15.2

Charging by Induction, 2

- A negatively charged rubber rod is brought near an uncharged sphere.
- The charges in the sphere are redistributed.
 - Some of the electrons in the sphere are repelled from the electrons in the rod.

Electrons redistribute when a charged rod is brought close.

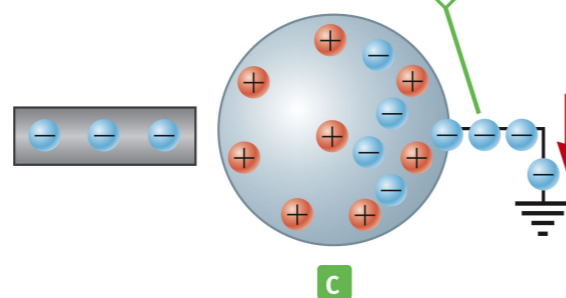


Section 15.2

Charging by Induction, 3

- The region of the sphere nearest the negatively charged rod has an excess of positive charge because of the migration of electrons away from this location.
- A grounded conducting wire is connected to the sphere.
 - Allows some of the electrons to move from the sphere to the ground

Some electrons leave the grounded sphere through the ground wire.

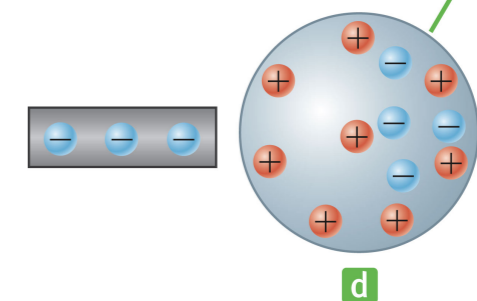


Section 15.2

Charging by Induction, 4

- The wire to ground is removed, the sphere is left with an excess of induced positive charge
- Initially, the positive charge on the sphere is nonuniformly distributed.

The excess positive charge is nonuniformly distributed.

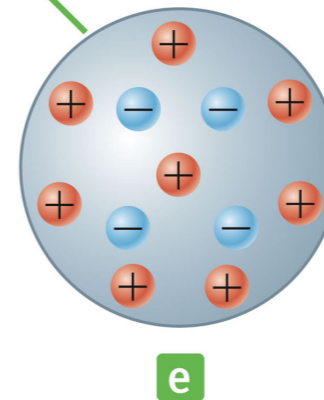


Section 15.2

Charging by Induction, Final

- Eventually, the excess positive charge becomes evenly distributed due to the repulsion between the positive charges.
- *Charging by induction requires no contact with the object inducing the charge.*

The remaining electrons redistribute uniformly, and there is a net uniform distribution of positive charge on the sphere's surface.



Charging results in polarization

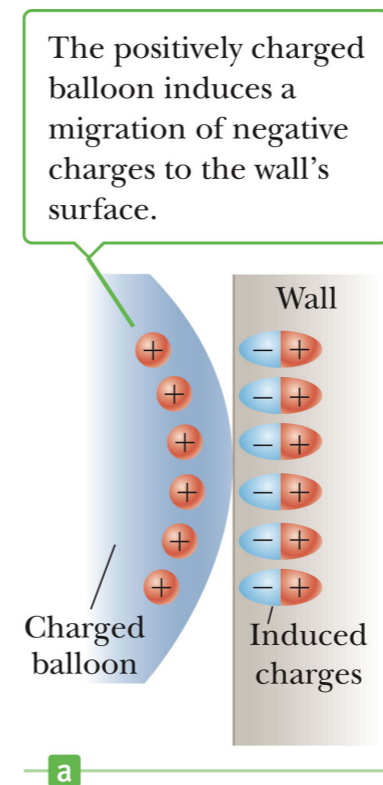
Polarization

- In most neutral atoms or molecules, the center of positive charge coincides with the center of negative charge.
- In the presence of a charged object, these centers may separate slightly.
 - This results in more positive charge on one side of the molecule than on the other side
- This realignment of charge on the surface of an insulator is known as *polarization*.

Polarization

Examples of Polarization

- The charged object (on the left) induces charge on the surface of the insulator.
- A charged comb attracts bits of paper due to polarization of the paper.



Question

1. Doug rubs a piece of fur on a hard rubber rod, giving the rod a negative charge. What happens?
 - a. Protons are removed from the rod.
 - b. Electrons are added to the rod.
 - c. The fur is also charged negatively.
 - d. The fur is left neutral.

Question

A repelling force must occur between two charged objects under which conditions?

- a. Charges are of unlike signs.
- b. Charges are of like signs.
- c. Charges are of equal magnitude.
- d. Charges are of unequal magnitude.

Question

4. A metallic object holds a charge of $-3.8 \times 10^6 \text{ C}$. What total number of electrons does this represent? ($e = 1.6 \times 10^{-19} \text{ C}$ is the magnitude of the electronic charge.)

- a. 4.2×10^{24}
- b. 6.1×10^{25}
- c. 2.4×10^{25}
- d. 1.6×10^{24}

Let's start getting quantitative

- * Coulomb's law
- * How charges attract or repel
- * Remember, this is somewhat contrived, the forces change as soon as motion starts

Coulomb's Law

- Coulomb shows that an electric force has the following properties:
 - It is directed along the line joining the two particles and inversely proportional to the square of the separation distance, r , between them
 - It is proportional to the product of the magnitudes of the charges, $|q_1|$ and $|q_2|$ on the two particles
 - It is attractive if the charges are of opposite signs and repulsive if the charges have the same signs

Coulomb's Law, Cont.

- Mathematically,

$$F = k_e \frac{|q_1| |q_2|}{r^2}$$

- k_e is called the *Coulomb Constant*
 - $k_e = 8.9875 \times 10^9 \text{ N m}^2/\text{C}^2$
- Typical charges can be in the μC range
 - Remember, Coulombs must be used in the equation
- Remember that force is a *vector* quantity
- Applies only to point charges and spherical distributions of charges
 - r is the distance between the two centers of charge

Characteristics of Particles

Table 15.1 Charge and Mass of the Electron, Proton, and Neutron

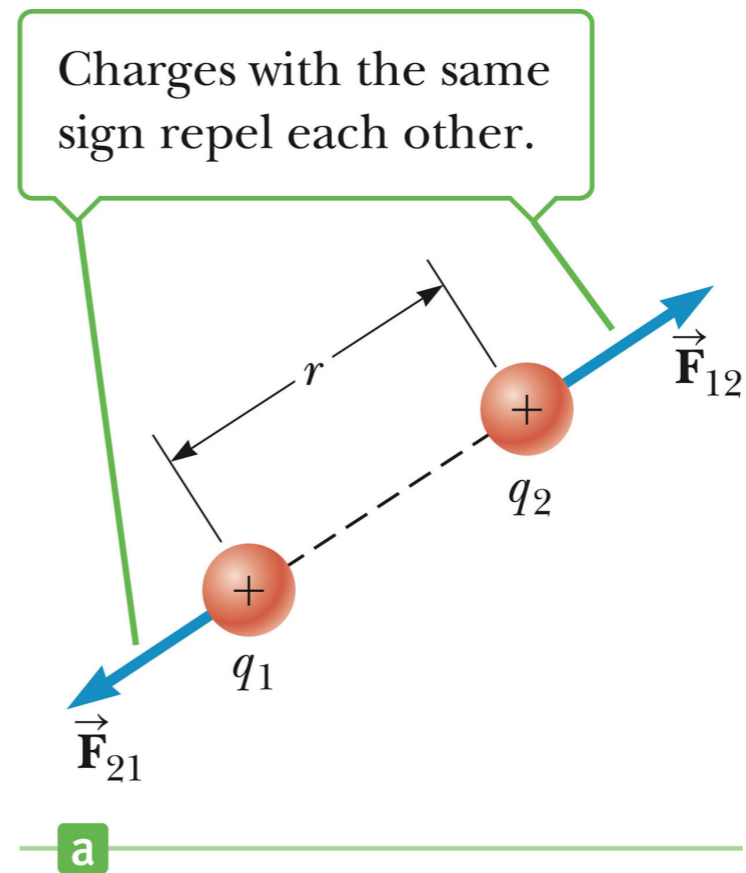
Particle	Charge (C)	Mass (kg)
Electron	-1.60×10^{-19}	9.11×10^{-31}
Proton	$+1.60 \times 10^{-19}$	1.67×10^{-27}
Neutron	0	1.67×10^{-27}

Caveat

- * Coulomb's law as we apply it is somewhat fictitious. In reality a force means an acceleration which means movement.
- * For our problems assume they mean all particles are held in place by a thumbtack.

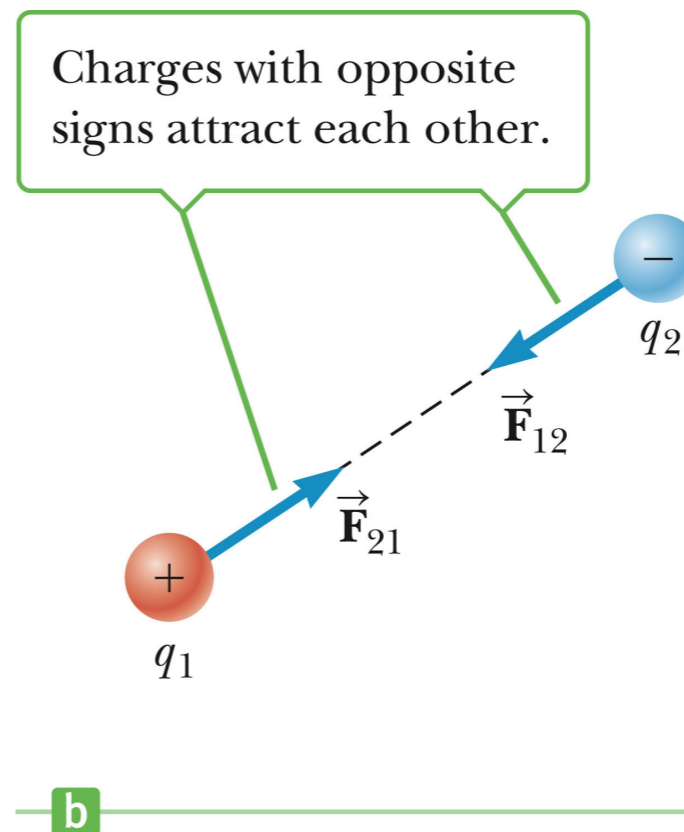
Vector Nature of Electric Forces

- Two point charges are separated by a distance r
- The like charges produce a repulsive force between them
- The force on q_1 is equal in magnitude and opposite in direction to the force on q_2



Vector Nature of Forces, Cont.

- Two point charges are separated by a distance r
- The unlike charges produce an attractive force between them
- The force on q_1 is equal in magnitude and opposite in direction to the force on q_2



Important

- * Calculate only the magnitude. Put the directions in by hand.

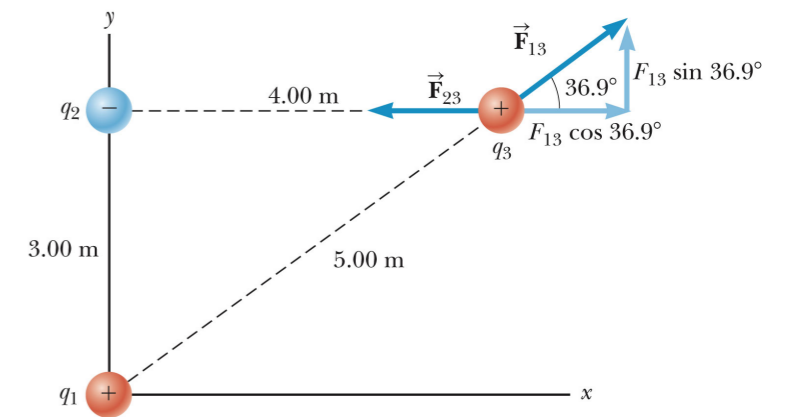
The Superposition Principle

- The resultant force on any one charge equals the vector sum of the forces exerted by the other individual charges that are present.
 - Find the electrical forces between pairs of charges separately
 - Then add the vectors
 - Remember to add the forces as *vectors*

Section 15.3

Superposition Principle Example

- The force exerted by q_1 on q_3 is \vec{F}_{13}
- The force exerted by q_2 on q_3 is \vec{F}_{23}
- The *total force* exerted on q_3 is the vector sum of \vec{F}_{13} and \vec{F}_{23}



Section 15.3

Let's do a couple
examples

The electric Field

- * Ok, a field is going to be a new concept for you guys
- * Here's how I like to think of a field (sort of)



A field

- * Is a convenient way of defining how a test particle would move placed in a region containing other particles that are exerting forces**

Electric Fields

- * Fields are represented by field lines
- * Protons act as sources for the field
- * Electrons as sinks. The directions of the arrows on the field line indicate which way a positive charge would move (reverse this for a negative charge).

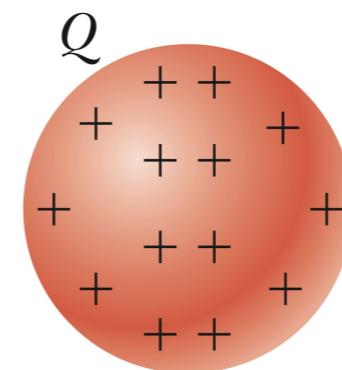
The Electric Field

Electrical Field

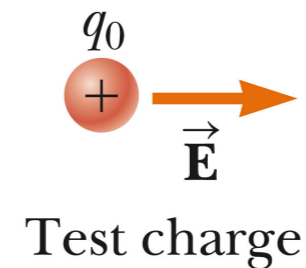
- Faraday developed an approach to discussing fields.
- An **electric field** is said to exist in the region of space around a charged object.
 - When another charged object enters this electric field, the field exerts a *force* on the second charged object.

Electric Field, Cont.

- A charged particle, with charge Q , produces an electric field in the region of space around it.
- A small *test charge*, q_0 , placed in the field, will experience a force.



Source charge



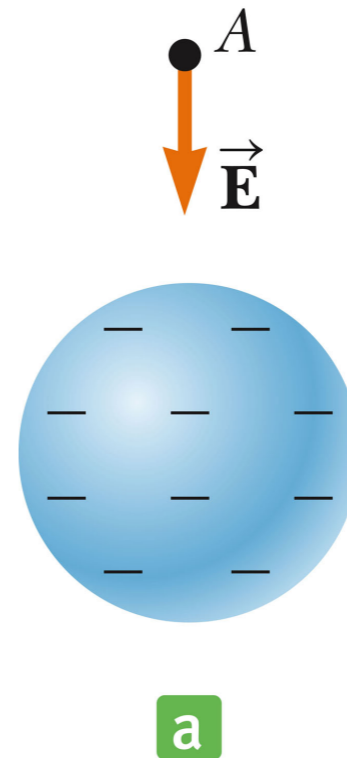
Test charge

Electric Field

- Mathematically, $\vec{E} = \frac{\vec{F}}{q_o} = \frac{k_e Q}{r^2}$
- SI unit: N / C
- Use this for the magnitude of the field
- The electric field is a vector quantity
- The direction of the field is defined to be the direction of the electric force that would be exerted on a small *positive* test charge placed at that point.

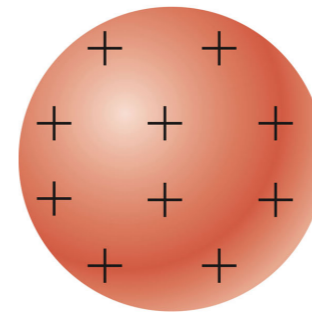
Direction of Electric Field

- The electric field produced by a negative charge is directed toward the charge.
 - A positive test charge would be attracted to the negative source charge.



Direction of Electric Field, Cont.

- The electric field produced by a positive charge is directed away from the charge.
 - A positive test charge would be repelled from the positive source charge.



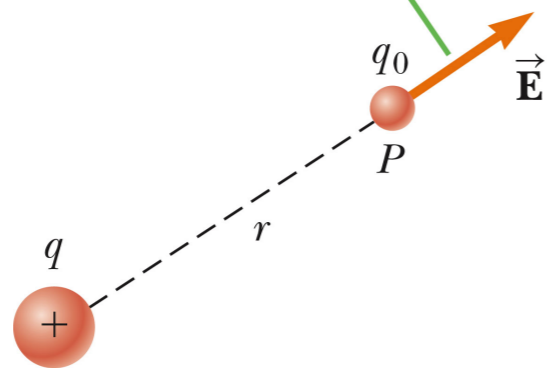
b

More About a Test Charge and The Electric Field

- The test charge is required to be a small charge.
 - It can cause no rearrangement of the charges on the source charge.
 - Mathematically, the size of the test charge makes no difference.
 - Using $q_o = 1 \text{ C}$ is convenient
- The electric field exists whether or not there is a test charge present.

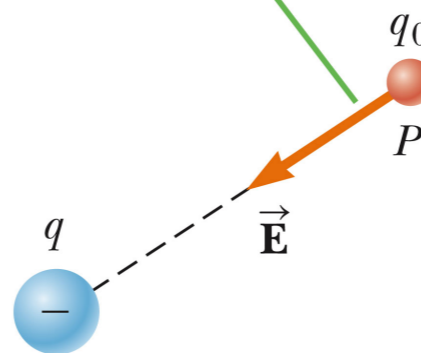
Electric Field, Direction Summary

If q is positive, the electric field at P points radially outwards from q .



a

If q is negative, the electric field at P points radially inwards toward q .



b

Electric Fields and Superposition Principle

- The superposition principle holds when calculating the electric field due to a group of charges.
 - Find the fields due to the individual charges.
 - Add them as vectors.
 - Use symmetry whenever possible to simplify the problem.

Problem Solving Strategy

- **Draw** a diagram of the charges in the problem.
- **Identify** the charge of interest.
 - You may want to circle it
- **Units** – Convert all units to SI.
 - Need to be consistent with k_e

Problem Solving Strategy, Cont.

- **Apply Coulomb's Law.**
 - For each charge, find the force on the charge of interest.
 - Determine the direction of the force.
 - The direction is always along the line of the two charges.
- **Sum all the x- and y- components.**
 - This gives the x- and y-components of the resultant force
- **Find the resultant force** by using the Pythagorean theorem and trigonometry.

Problem Solving Strategy, Electric Fields

- Calculate Electric Fields of point charges.
 - Use the equation to find the electric field due to the individual charges.
 - The direction is given by the direction of the force on a positive test charge.
 - The Superposition Principle can be applied if more than one charge is present.

Electric Field Lines

- A convenient aid for visualizing electric field patterns is to draw lines pointing in the direction of the field vector at any point.
- These lines are called **electric field lines** and were introduced by Michael Faraday.

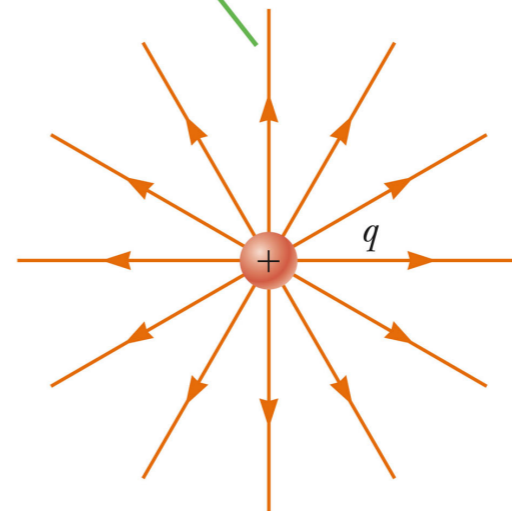
Electric Field Lines, Cont.

- The field lines are related to the field in the following manners:
 - The electric field vector \vec{E} , is tangent to the electric field lines at each point.
 - The number of lines per unit area through a surface perpendicular to the lines is proportional to the strength of the electric field in a given region.

Electric Field Line Patterns

- Point charge
- The lines radiate equally in all directions.
- For a positive source charge, the lines will radiate outward.

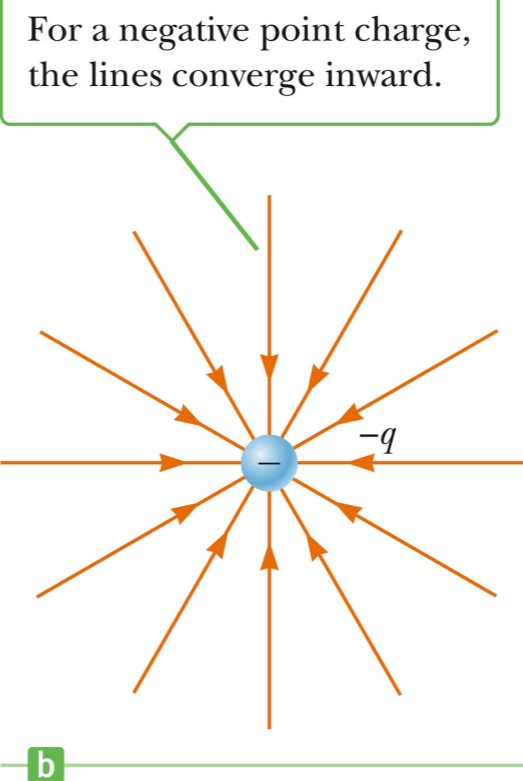
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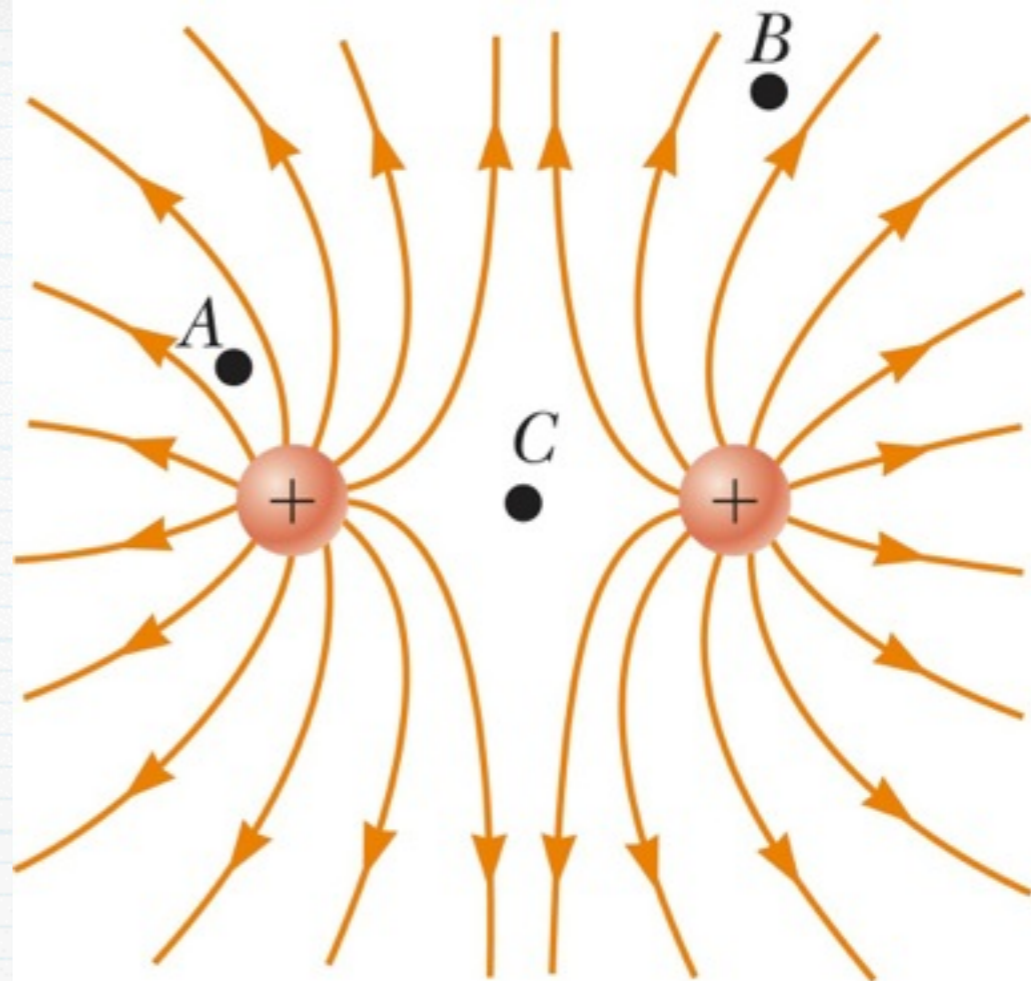
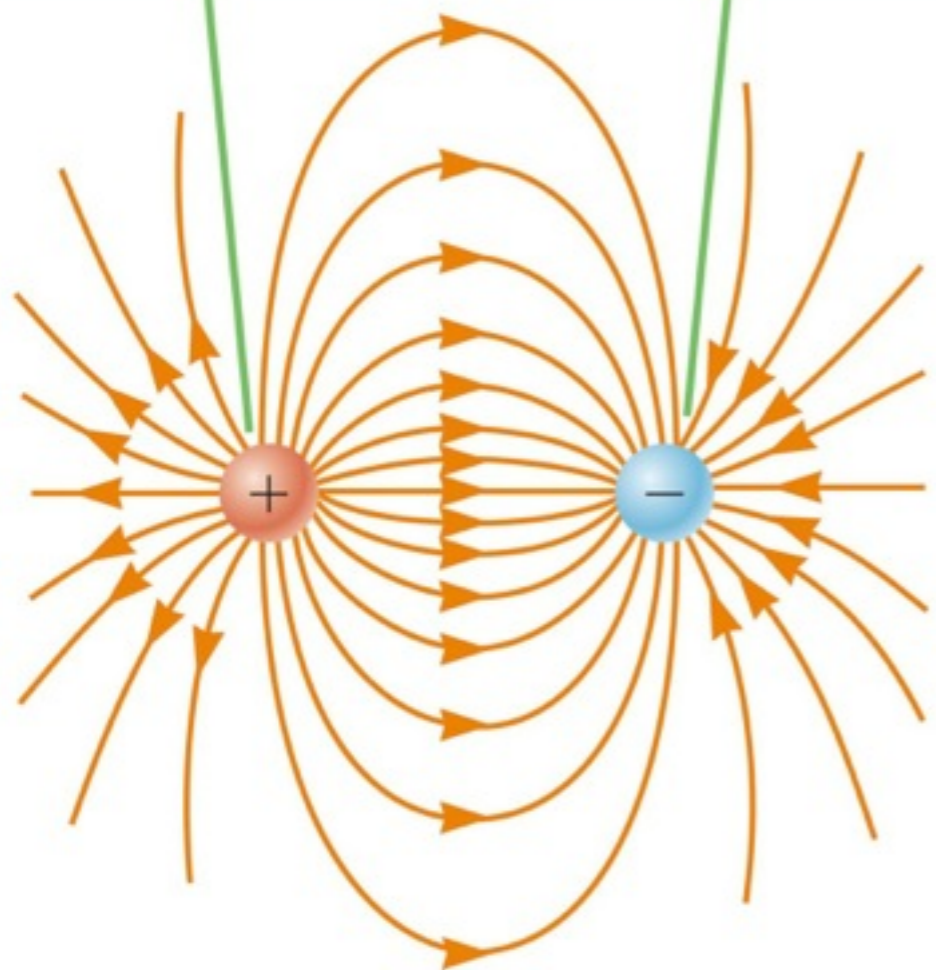
a

Electric Field Line Patterns

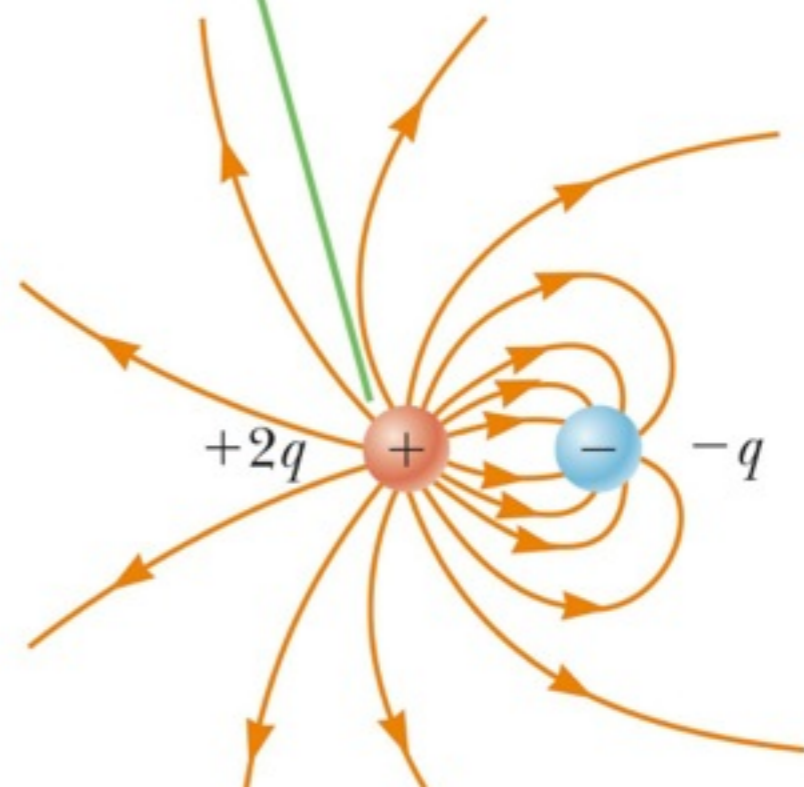
- For a negative source charge, the lines will point inward.



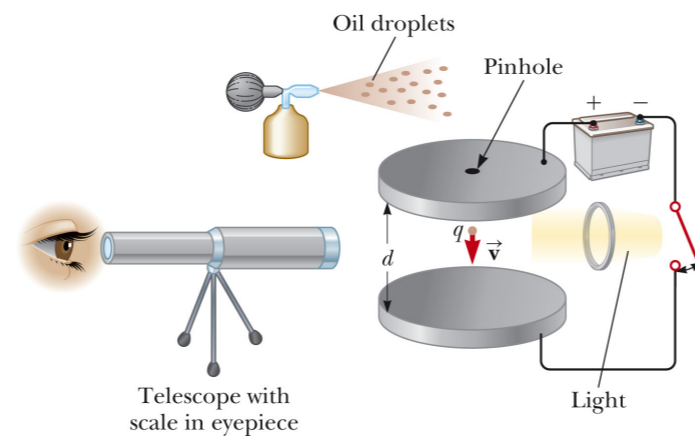
The number of field lines leaving the positive charge equals the number terminating at the negative charge.



Two field lines leave $+2q$ for every one that terminates on $-q$.

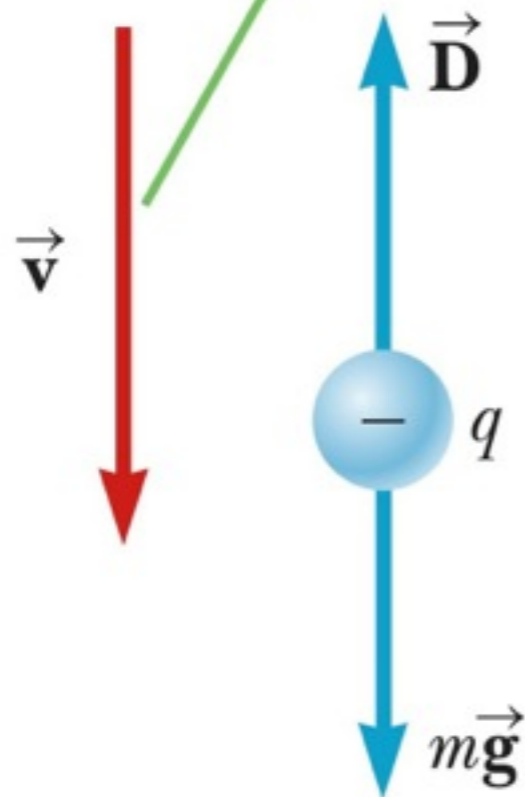


Millikan Oil-Drop Experiment



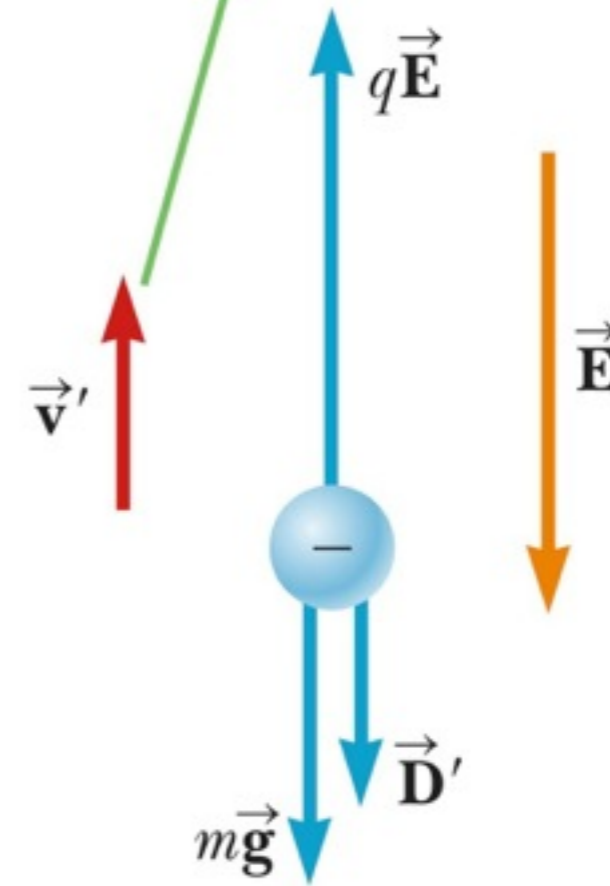
- Measured the elementary charge, e
 - Found every charge had an integral multiple of e
- $$-q = n e$$

Electric field off: the droplet falls at terminal velocity \vec{v} , the gravity and drag forces summing to zero.



a

Electric field on: the droplet moves upward at new terminal velocity \vec{v}' , the gravity, drag, and electric forces summing to zero.



b

Question

If the distance between two point charges is tripled, the mutual force between them will be changed by what factor?

- a. 9.0
- b. 3.0
- c. 0.33
- d. $1/9$

Question

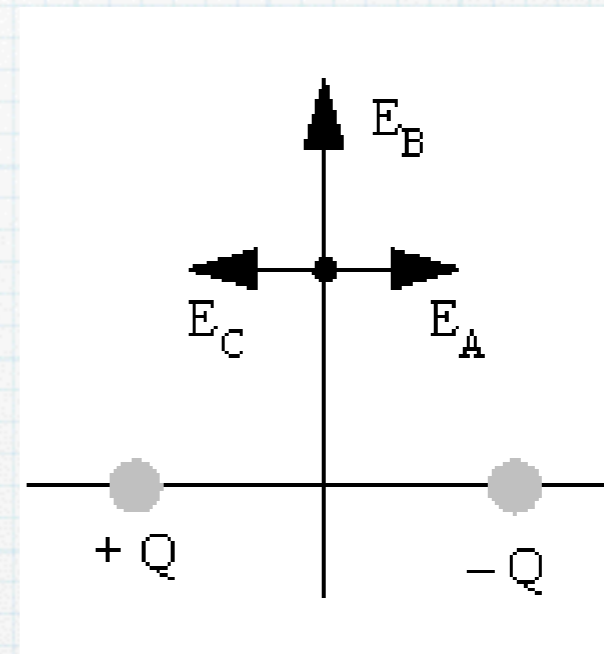
An electron with a charge value of $1.6 \times 10^{-19} \text{ C}$ is moving in the presence of an electric field of 400 N/C . What force does the electron experience?

- a. $2.3 \times 10^{-22} \text{ N}$
- b. $1.9 \times 10^{-21} \text{ N}$
- c. $6.4 \times 10^{-17} \text{ N}$
- d. $4.9 \times 10^{-17} \text{ N}$

Question

Two charges, $+Q$ and $-Q$, are located two meters apart and there is a point along the line that is equidistant from the two charges as indicated. Which vector best represents the direction of the electric field at that point?

- a. Vector E_A
- b. Vector E_B
- c. Vector E_C
- d. The electric field at that point is zero.



Let's do some examples

Key Concepts

- * Charge is conserved
- * Charge is quantized $e = 1.6 \times 10^{-19} \text{ C}$
- * Only NET charge matters
- * Charging may be through conduction or induction
- * Opposites attract, likes repel
- * Electric forces and fields are VECTOR quantities
- * Electric forces and fields obey the superposition principle

Key Equations

$$\vec{F}_e = k \frac{q_1 q_2}{r^2} \hat{r}$$

$$\vec{E} = k \frac{q}{r^2} \hat{r} = \frac{\vec{F}_e}{q_0}$$