

## Modern Physics: Homework 2

Due: 2 February 2026

### 1 Photon absorption

An atom initially at rest absorbs a photon with wavelength 500 nm and subsequently emits a photon with wavelength 660 nm.

- a) Determine the energy gained by the atom in this process.
- b) Suppose that the atom has mass  $3.3 \times 10^{-26}$  kg. Determine the speed of the atom after it has emitted the photon, assuming that any additional energy it acquired only contributed to its motion.

### 2 Down conversion

In “down conversion” a photon with wavelength  $\lambda_i$  is incident on a crystal. The crystal absorbs this and emits two photons. The crystal’s energy after this is the same as before the photon absorption. The wavelengths of the emitted photons are  $\lambda_{e1}$  and  $\lambda_{e2}$  and these are not necessarily equal.

- a) Which of the following is true?
  - i) Both  $\lambda_{e1}$  and  $\lambda_{e2}$  are larger than  $\lambda_i$ .
  - ii) Both  $\lambda_{e1}$  and  $\lambda_{e2}$  are smaller than  $\lambda_i$ .
  - iii) One of  $\lambda_{e1}$  and  $\lambda_{e2}$  is larger than  $\lambda_i$ , the other is smaller.

Explain your choice.

- b) Determine an expression that relates  $\lambda_{e1}$ ,  $\lambda_{e2}$  and  $\lambda_i$ .
- c) Light with wavelength 415 nm is incident on such a crystal and two photons with identical wavelength are emitted. Determine the wavelength of the emitted photons.

### 3 Compton Effect

In the original Compton effect experiment the wavelength of the incident radiation used was 0.0711 nm.

- a) Determine the wavelength shift for scattering from an electron at  $20^\circ$ ,  $50^\circ$  and  $90^\circ$ .
- b) Suppose that the same radiation was incident on a hydrogen atom. Would the wavelength shift be more difficult or less difficult to observe than for scattering from an electron? Explain your answer.
- c) Suppose that visible light were used in an attempt to observe the Compton effect off an electron. Would this be more or less difficult to observe than for the 0.0711 nm radiation? Explain your answer.

#### 4 Compton Effect and Mass of Scattering Object

The Compton effect is usually observed by bombarding electrons with X-rays or gamma rays. For example, Compton used X-rays with wavelength 0.0711 nm. The change in wavelength will depend on the mass of the object from which the radiation is scattered. This enters into the analysis via the Compton wavelength  $\lambda_c = h/mc$  where  $m$  is the mass of the scattering object.

- a) For a scattering object of mass  $m$  find an expression for the maximum change in wavelength (as the scattering angle varies) between the incident photon and scattered photon. Express this in terms of the Compton wavelength of the scattering object.
- b) Find an expression for the maximum fractional change in the wavelength, i.e.  $(\lambda_2 - \lambda_1)/\lambda_1$ , as a function  $\lambda_1$  and the Compton wavelength.
- c) Calculate the Compton wavelength for an electron and determine the maximum fractional change in wavelength for the X-rays used by Compton.
- d) Calculate the Compton wavelength for a proton and determine the maximum fractional change in wavelength for the X-rays used by Compton.
- e) Calculate the Compton wavelength for an object of mass 1 kg and determine the maximum fractional change in wavelength for the X-rays used by Compton. What does this tell you about the possibility of detecting Compton scattering from everyday objects using X-rays?

#### 5 Mathematical representation of waves

Consider a wave that is represented by

$$\Psi(x, t) = A \cos(kx - \omega t)$$

where  $k = 2\pi/\lambda$  and  $\omega = 2\pi f$ . The aim of the following exercises is to show that this expression captures many of the intuitive features of waves.

- a) Consider a snapshot of the wave at  $t = 0$ . Use the expression to find the possible values of  $x$  at which the crests (maximum points) of the wave are located. Determine an expression (in terms of parameters that appear in this problem) for the distance between successive crests.
- b) Show that the expression predicts that at any time  $t$  the value of  $\psi$  is the same for any two points separated by exactly one wavelength, i.e. show that

$$\Psi(x + \lambda, t) = \Psi(x, t).$$

- c) As time passes one can follow a particular crest by focusing on the argument of the cosine. For example suppose that  $kx - \omega t = 0$ . This describes one particular crest of  $\psi$ . Where is this crest located at time  $t_i = 0$ ? Determine an expression for the location of this crest located at any later time  $t_f$ . How far has the crest traveled during the time interval from  $t_i$  to  $t_f$ ? Use this to determine an expression (in terms of  $k$  and  $\omega$ ) for the speed with which this crest travels.

## 6 Quantum physics article

The [Physics magazine website](#) provides articles that summarize recent highlights of quantum physics. Visit the website and scan it for an article that interests you. When you use this website either “Quantum Physics” or “Quantum Information” boxes must be checked under “Subject Area.”

The idea is for you to summarize the article as best as you can. Some of these articles will contain technical details that might be difficult to understand. It’s less important for you to understand the article than to get an overall idea of what it tries to describe.

- a) Provide the article title.
- b) Briefly describe, as best as you can, the contents of the article.
- c) Briefly describe why you find the article interesting or why you think it is important.
- d) Go to D2L. In the Quantum Physics Highlight Articles discussion forum, find the Quantum Physics Highlight: HW 2 topic. Provide a link to your article there and a one sentence description of why anyone should read it (everyone in the class will eventually see this).