

Fri: HW by Spm

Photon description of light

The photon picture of light is as follows

A diagram showing a light source on the left emitting individual photons on the right. The photons are represented by dashed lines with arrows pointing away from the source.

light source individual photons

Light source emits individual photons. Want to predict the probability with which they arrive at any location

TO CALCULATE

Associate a wave with light

The wave is described by wavelength, frequency and intensity

A diagram showing a cylindrical light source on the left emitting a wave on the right. The wave is represented by several curved lines.

wave associated with photons (their bookkeeping device)

Probability of arrival on screen / detectors

A graph showing a red curve representing the probability of arrival on a screen. The x-axis is labeled 'screen' and the y-axis is labeled 'probability'.

probability

screen

A diagram showing light rays hitting a screen. The light is represented by several parallel lines, and the screen is a vertical line.

light

Use classical physics of waves to predict intensity

A graph showing a green curve representing intensity on a screen. The x-axis is labeled 'screen' and the y-axis is labeled 'intensity'.

intensity

screen

A diagram showing waves hitting a screen. The waves are represented by several curved lines, and the screen is a vertical line.

waves

Intensity profile gives probability distribution

MATHEMATICAL PICTURE

Slide 1

Quiz 1 100%

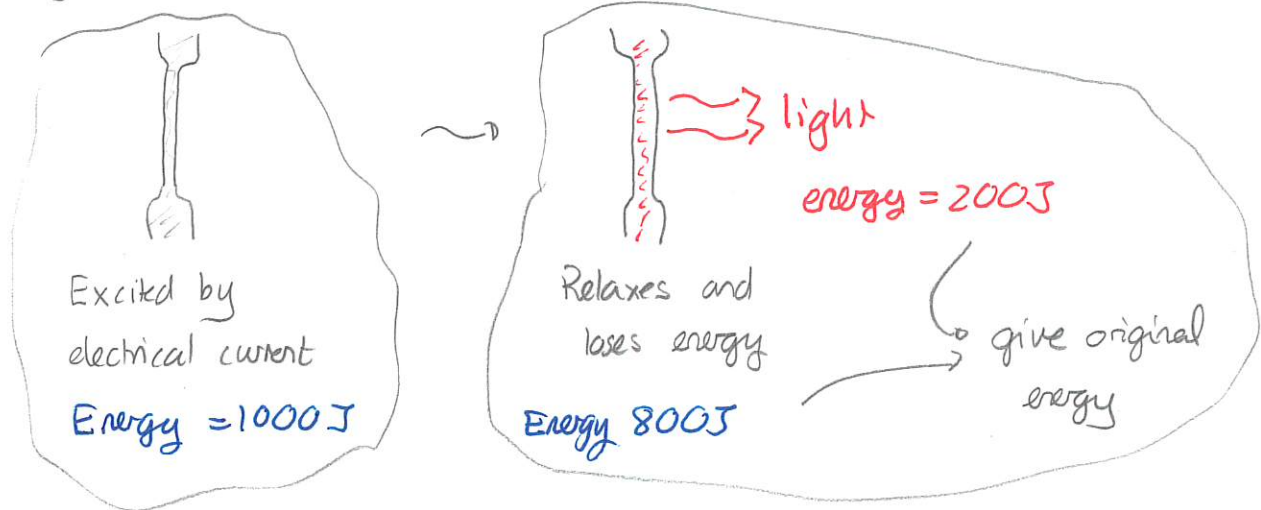
Quiz 2 50%

Photon energy

We need to describe the interactions of light and matter in terms of photons. The key ingredient will be energy. For example we can get a gas to emit light

DEMO: Spectrum tube with neon

The way that this works is that an electrical current excites the gas molecules. They then relax, losing energy.



In general we expect that for light of one particular type, every photon will be identical and therefore have the same energy. We will see that:

Consider light of one particular frequency. Each photon of this light has the same energy as any other photon of this light.

The energy of each photon will be different to that of a light source with a different frequency.

Note that frequency is associated with color. We now explore this for photons produced by different sources outside the visible spectrum.

Group Exercise 5

1 Photon energies and numbers, coin analogy

A particular light source can produce photons, each with energy 25×10^{-20} J. Consider a pulse of light produced by this source. Remember that the pulse consists of a number of individual photons.

- Consider the total energy in the pulse. List the five lowest possible total energies that the pulse could have.
- Is it possible that the pulse has total energy 175×10^{-20} J? Explain your answer.
- Is it possible that the pulse has total energy 185×10^{-20} J? Explain your answer.

There is an analogy with money. In the following, suppose that the only cash one has is a collection of coins and that the only coins in the collection are quarters.

- List the five lowest possible amounts of cash that one could have in the collection.
- Is it possible that the amount of cash one has is \$1.75? Explain your answer.
- Is it possible that the amount of cash one has is \$1.85? Explain your answer.
- Suppose that the amount of cash is \$18.50. Determine the number of coins in this collection. How did you do this?
- Now suppose that the collection of coins only consisted of dimes. List the five lowest possible amounts of cash in the collection. If the total amount of cash were \$18.50, how many coins would the collection contain? Which collection contains more coins? How did you determine this?

Now consider photons produced by different light sources. One source produces photons, each with energy 25×10^{-20} J. The other source produces photons, each with energy 10×10^{-20} J.

- Suppose that each source produces light with total energy 1.75×10^{-18} J = 175×10^{-20} J. Determine the number of photons produced by each light source. Which requires more photons?

Answer a)

Number of photons	Energy
0	0 J
1	25×10^{-20} J
2	50×10^{-20} J
3	75×10^{-20} J
4	100×10^{-20} J
5	125×10^{-20} J

b) Yes, continuing the list would give 7 photons $\rightarrow 175 \times 10^{-20}$ J

c) No, with eight photons $\rightarrow 200 \times 10^{-20}$ J

No whole number of ^{these} photons would give 185×10^{-20} J.

d)

Number of coins	Cash
0	\$ 0 = 0c
1	25c
2	50c
3	75c
4	100c = \$1.00
5	125c = \$1.25

e) Yes 7 quarters gives 175c = \$1.75

f) No. 8 quarters would give 200c = \$2.00. The number \$1.85 would require between 7 and 8 quarters. This is not possible.

g) number of quarters = $\frac{\text{total value of cash}}{\text{value of one quarter}}$

$$= \frac{\$ 18.50}{\$ 0.25} = 74$$

Divide total value by value of one coin.

h)

Number of coins	Cash
0	0
1	10c
2	20c
3	30c
4	40c
5	50c

number of dimes = $\frac{\$ 18.50}{\$ 0.10} = 185$ dimes.

The collection of dimes contains more coins.

i) Source with $25 \times 10^{-20} \text{ J}$ photons requires $\frac{150 \times 10^{-20} \text{ J}}{25 \times 10^{-20} \text{ J}} = 6$ photons

Source with $10 \times 10^{-20} \text{ J}$ photons requires $\frac{150 \times 10^{-20} \text{ J}}{10 \times 10^{-20} \text{ J}} = 15$ photons.

We need a rule for the energy of any photon. This clearly depends on the source of light. The rule, proposed by Einstein in 1905 is:

The energy of a single photon of light only depends on the frequency of the wave associated with the light. The energy of one photon is:

$$\text{energy} = h \times \text{frequency} \quad E = h \times f.$$

where

in Hertz

$$h = 6.63 \times 10^{-34} \text{ J}\cdot\text{s}$$

is Planck's constant

Note that for light we can determine frequency from wavelength using

$$\text{speed of light} = \text{frequency} \times \text{wavelength}$$

$$\Rightarrow \text{frequency} = \frac{\text{speed of light}}{\text{wavelength}}$$

Quiz 3 90%

For a given wavelength of light, the energy of every photon is the same.