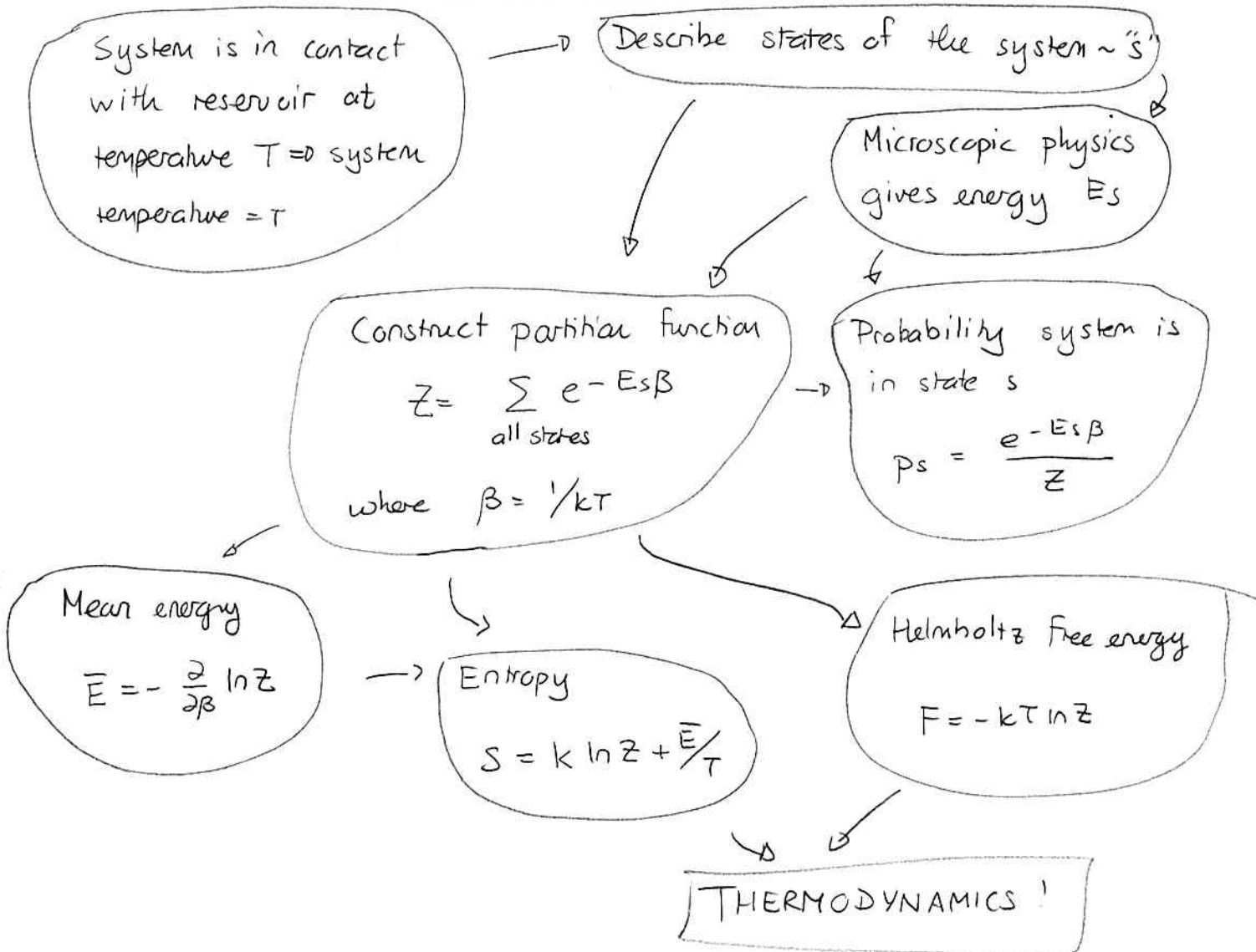


Mon: HW by 6pm

Tues: Read 4.8, 6.1

Canonical ensemble

The framework for the canonical ensemble is:



We briefly consider the entropy and Helmholtz free energy statements.

Entropy and Helmholtz Free Energy

The entropy is

$$\begin{aligned} S &= -k \sum p_s \ln[p_s] \\ &= -k \sum \frac{e^{-E_s \beta}}{Z} \ln \left[\frac{e^{-E_s \beta}}{Z} \right] \\ &= -k \sum \frac{e^{-E_s \beta}}{Z} \left\{ \ln[e^{-E_s \beta}] - \ln Z \right\} \\ &= -k \sum \frac{e^{-E_s \beta}}{Z} (-E_s \beta) + k \frac{\ln Z}{Z} \underbrace{\sum e^{-E_s \beta}}_Z \\ &= k \beta \underbrace{\sum E_s \frac{e^{-E_s \beta}}{Z}}_{\bar{E}} + k \ln Z \\ &= \frac{k}{kT} \bar{E} + k \ln Z \end{aligned}$$

Thus

If we identify the thermodynamic energy as $E = \bar{E}$ then

$$S = k \ln Z + E/T$$

Separately $F = E - TS$ gives

The Helmholtz free energy is

$$F = -kT \ln Z$$

1
 § Mean value of energy for a single one-dimensional quantum oscillator

A one dimensional quantum oscillator has energy states labeled by $n = 0, 1, 2, \dots$ with energy $E_n = \hbar\omega (n + 1/2)$.

- Determine the partition function for the oscillator.
- Determine the mean value of the energy for the oscillator.
- Determine the uncertainty in the oscillator energy.
- Suppose the $\hbar\omega \gg kT$. Determine an expression for the mean value of the energy for the oscillator. What does this imply about the state of the oscillators in the ensemble?
- Suppose the $\hbar\omega \ll kT$. Determine an expression for the mean value of the energy for the oscillator.

Answer: a) List states and energies

label	n	Energy E_n
	0	$\hbar\omega/2$
	1	$3\hbar\omega/2$
	2	$5\hbar\omega/2$
	\vdots	
	n	$\hbar\omega(n + 1/2)$
	\vdots	

Then

$$\begin{aligned}
 Z &= \sum_{n=0}^{\infty} e^{-E_n \beta} \\
 &= \sum_{n=0}^{\infty} e^{-\hbar\omega(n+1/2)\beta} \\
 &= e^{-\hbar\omega\beta/2} \sum_{n=0}^{\infty} e^{-\hbar\omega\beta n}
 \end{aligned}$$

The latter is a geometric series with $x = e^{-\hbar\omega\beta}$

So $1+x+x^2+\dots = \frac{1}{1-x}$ provided $x < 1$.

In this case $x = e^{-\hbar\omega\beta} < 1$ and thus

$$Z = \frac{e^{-\hbar\omega\beta/2}}{1 - e^{-\hbar\omega\beta}}$$

$$b) \quad \bar{E} = -\frac{\partial}{\partial\beta} \ln Z = -\frac{\partial}{\partial\beta} \ln \left[\frac{e^{-\hbar\omega\beta/2}}{1 - e^{-\hbar\omega\beta}} \right]$$

$$= -\frac{\partial}{\partial\beta} \left\{ \ln[e^{-\hbar\omega\beta/2}] - \ln[1 - e^{-\hbar\omega\beta}] \right\}$$

$$= -\frac{\partial}{\partial\beta} \left\{ -\hbar\omega\beta/2 - \ln(1 - e^{-\hbar\omega\beta}) \right\}$$

$$= \frac{\hbar\omega}{2} + \frac{\partial}{\partial\beta} \ln(1 - e^{-\hbar\omega\beta})$$

$$= \frac{\hbar\omega}{2} + \frac{e^{-\hbar\omega\beta}}{1 - e^{-\hbar\omega\beta}} \hbar\omega$$

$$\Rightarrow \bar{E} = \hbar\omega \left\{ \frac{1}{e^{\hbar\omega\beta} - 1} + \frac{1}{2} \right\}$$

$$c) \quad \sigma_E^2 = -\frac{\partial \bar{E}}{\partial\beta} = -\hbar\omega \frac{-\hbar\omega e^{\hbar\omega\beta}}{(e^{\hbar\omega\beta} - 1)^2}$$

$$= (\hbar\omega)^2 \frac{e^{\hbar\omega\beta}}{(e^{\hbar\omega\beta} - 1)^2}$$

$$\Rightarrow \sigma_E = \hbar\omega \frac{e^{\hbar\omega\beta/2}}{e^{\hbar\omega\beta} - 1}$$

$$d) \quad \hbar\omega \gg kT \Rightarrow \frac{\hbar\omega}{kT} \gg 1 \Rightarrow \hbar\omega\beta \gg 1$$

$$\Rightarrow e^{\hbar\omega\beta - 1} \approx e^{\hbar\omega\beta}$$

$$\text{Thus } \bar{E} \approx \hbar\omega \left\{ e^{-\hbar\omega\beta} + \frac{1}{2} \right\} \Rightarrow \bar{E} \approx \hbar\omega/2$$

\uparrow
negligible

The typical oscillator must be in the ground state.

$$e) \quad \hbar\omega\beta \ll 1 \quad e^{\hbar\omega\beta} \approx 1 + \hbar\omega\beta$$

$$\Rightarrow \bar{E} \approx \hbar\omega \left\{ \frac{1}{\hbar\omega\beta} + \frac{1}{2} \right\}$$

$$\approx \frac{\hbar\omega}{\hbar\omega\beta} = kT \Rightarrow \bar{E} = kT.$$

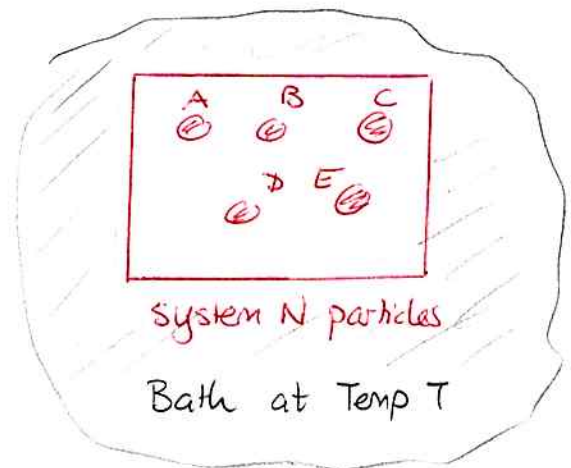
Multiple particles

The canonical ensemble formalism applies equally well to systems that contain a single particle or many particles. In general thermodynamic ensembles consist of many particles, typically identical and often non-interacting. The

thermodynamics can be generated by finding the partition function for the entire system.

Let

Z = partition function for entire system of all particles



We could separately consider a system consisting only of particle A and we could construct

Z_A = partition function for system only consisting of A

Z_B = " " " " " " " " B

etc... Each of these would be easier to compute than Z . So we now consider the possibility:

Given partition functions
 Z_A, Z_B, Z_C, \dots

→ Can one easily construct
 Z from Z_A, Z_B, \dots ?

In general we require that A, B, C, D, ... be distinguishable (this only generally applies in classical physics).

Then

If systems A, B, C, D, \dots are distinguishable and non-interacting

$$Z = Z_A Z_B Z_C \dots$$

where Z is the partition function for the collection A, B, C, D, \dots

and Z_A is the partition function for a system just consisting of A , etc...

Proof: Let

$\{S_A\}$ be the states for A S_A has energy E_{S_A}

$\{S_B\}$ " " " " B S_B " " E_{S_B}

Then the states for the ensemble are collections:

(S_A, S_B, S_C, \dots) \rightsquigarrow energy $E_{S_A} + E_{S_B} + E_{S_C} + \dots$

no interaction terms.

$$\text{So } Z = \sum_{S_A, S_B, S_C} e^{-[E_{S_A} + E_{S_B} + E_{S_C} + \dots] \beta}$$

$$= \sum_{S_A, S_B, S_C} e^{-E_{S_A} \beta} e^{-E_{S_B} \beta} \dots$$

$$= \underbrace{\sum_{S_A} e^{-E_{S_A} \beta}}_{Z_A} \underbrace{\sum_{S_B} e^{-E_{S_B} \beta}}_{Z_B} \dots$$

$$\Rightarrow Z = Z_A Z_B \dots$$

2 Pair of distinguishable particles

Consider a system that consists of two identical distinguishable particles. Each particle can be in one of three states, the energies for which are $0, \epsilon, 2\epsilon$.

- Considering the pair of particles as "the system," list all possible states of the system and their energies.
- Determine the partition function of the two-particle system.
- Consider just one of the particles as "the system." Determine the partition function for this single particle system.
- Show that the partition function for the two particle system is the product of the partition functions for the single particle system.

Answer: a) Label the particles A and B. Then

State A	state B	energy
0	0	0
0	1	ϵ
0	2	2ϵ
1	0	ϵ
1	1	2ϵ
1	2	3ϵ
2	0	2ϵ
2	1	3ϵ
2	2	4ϵ

$$\begin{aligned} \text{b) } Z_{two} &= \sum_s e^{-E_s \beta} \\ &= e^{-0\beta} + e^{-\epsilon\beta} + e^{-2\epsilon\beta} + e^{-\epsilon\beta} + e^{-2\epsilon\beta} + e^{-3\epsilon\beta} \\ &\quad + e^{-2\epsilon\beta} + e^{-3\epsilon\beta} + e^{-4\epsilon\beta} \\ &= 1 + 2e^{-\epsilon\beta} + 3e^{-2\epsilon\beta} + 2e^{-3\epsilon\beta} + e^{-4\epsilon\beta} \end{aligned}$$

c) For a single particle

$$\begin{aligned} Z_{\text{one}} &= \sum e^{-E_s \beta} \\ &= e^{-0\beta} + e^{-\epsilon\beta} + e^{-2\epsilon\beta} \\ &= 1 + e^{-\epsilon\beta} + e^{-2\epsilon\beta} \end{aligned}$$

$$\begin{aligned} \text{d) } Z_{\text{one}} Z_{\text{one}} &= (1 + e^{-\epsilon\beta} + e^{-2\epsilon\beta})(1 + e^{-\epsilon\beta} + e^{-2\epsilon\beta}) \\ &= 1 + e^{-\epsilon\beta} + e^{-2\epsilon\beta} + e^{-\epsilon\beta} + e^{-2\epsilon\beta} + e^{-3\epsilon\beta} \\ &\quad + e^{-2\epsilon\beta} + e^{-3\epsilon\beta} + e^{-4\epsilon\beta} \\ &= 1 + 2e^{-\epsilon\beta} + 3e^{-2\epsilon\beta} + 2e^{-3\epsilon\beta} + e^{-4\epsilon\beta} \\ &= Z_{\text{two}} \end{aligned}$$

$$\Rightarrow Z_{\text{two}} = Z_{\text{one}} Z_{\text{one}}$$

Energy for multiple non-interaction distinguishable particles

We can immediately reach conclusions about the energy for systems on non-interacting distinguishable particles.

$$\bar{E} = - \frac{\partial}{\partial \beta} \ln Z$$

$$= - \frac{\partial}{\partial \beta} \ln [Z_A Z_B Z_C \dots]$$

$$= - \frac{\partial}{\partial \beta} \left\{ \ln Z_A + \ln Z_B + \ln Z_C + \dots \right\}$$

$$= \underbrace{- \frac{\partial}{\partial \beta} \ln Z_A}_{\bar{E}_A} - \underbrace{\frac{\partial}{\partial \beta} \ln Z_B}_{\bar{E}_B} - \underbrace{\frac{\partial}{\partial \beta} \ln Z_C}_{\bar{E}_C}$$

$$\Rightarrow \bar{E} = \bar{E}_A + \bar{E}_B + \bar{E}_C$$

So the energy is additive. Then consider the uncertainty in the energy.

$$\sigma_E^2 = - \frac{\partial \bar{E}}{\partial \beta} = - \frac{\partial \bar{E}_A}{\partial \beta} - \frac{\partial \bar{E}_B}{\partial \beta} - \frac{\partial \bar{E}_C}{\partial \beta}$$

$$= \sigma_{E_A}^2 + \sigma_{E_B}^2 + \sigma_{E_C}^2$$

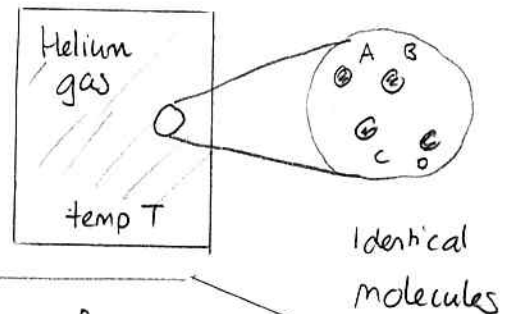
$$\Rightarrow \sigma_E = \sqrt{\sigma_{E_A}^2 + \sigma_{E_B}^2 + \sigma_{E_C}^2 + \dots}$$

This is not additive because of the square root.

Thermodynamic ensembles

Generally thermodynamic ensembles consist of a large number of identical systems. Consider an ensemble consisting of N identical distinguishable, non-interacting systems.

Then the previous more general results immediately give:



The partition function for an ensemble of N identical, distinguishable, non-interacting systems is

$$Z_{\text{ensemble}} = (Z_{\text{single particle}})^N$$

where

$$Z_{\text{single particle}} = \sum_s e^{-E_s \beta}$$

$s = \text{single particle states}$

Then it immediately follows that

For an ensemble of N identical, distinguishable, non-interacting particles

$$\bar{E}_{\text{ensemble}} = N \bar{E}_{\text{single particle}}$$

and

$$\sigma_{E_{\text{ensemble}}} = \sqrt{N} \sigma_{E_{\text{single particle}}}$$

These can give statistical quantities such as

mean energy $\bar{E} = \bar{E}(N, T)$

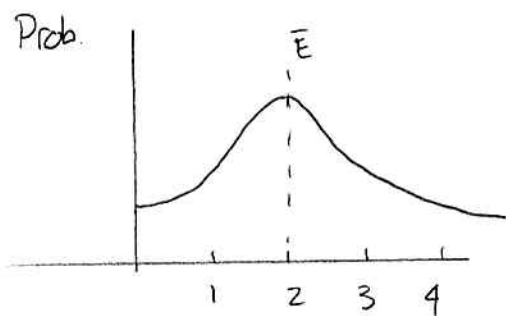
mean entropy $\bar{S} = \bar{S}(N, T)$

We make the identification

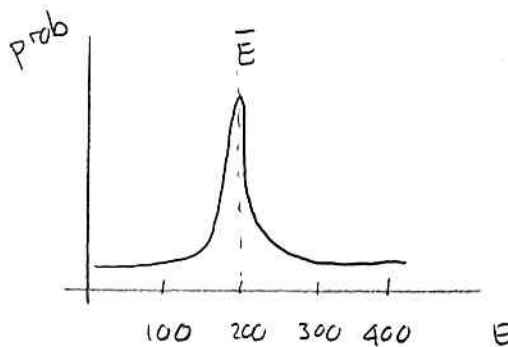
statistical physics $\leadsto \bar{E} = E$ for thermodynamics

Now statistical physics will give fluctuations in \bar{E} and we do not want such fluctuations to affect E that is used in thermodynamics.

We note though that as N increases the relative size of the fluctuations decreases. Roughly



one particle



100 particles

Generally as N increases the probability distribution (energy per particle) decreases and \bar{E} becomes more representative of the energy. We can compare the typical size of the fluctuations to the typical energy:

$$\frac{\sigma_E}{\bar{E}} \Big|_{N \text{ particles}} = \frac{\sqrt{N} \sigma_{E \text{ single}}}{N \bar{E} \text{ single}} = \frac{1}{\sqrt{N}} \frac{\sigma_{E \text{ single}}}{\bar{E} \text{ single}} \rightarrow 0 \text{ as } N \rightarrow \infty$$

As $N \rightarrow \infty$ the distribution of possible energies is only appreciably non-zero over a relatively vanishing range of energies centered on \bar{E} . Thus \bar{E} accurately represents the ensemble energy as $N \rightarrow \infty$